Notes

Selective Removal of an N-BOC Protecting Group in the Presence of a *tert*-Butyl Ester and Other Acid-Sensitive Groups

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Received December 13, 1993 (Revised Manuscript Received March 8, 1994)

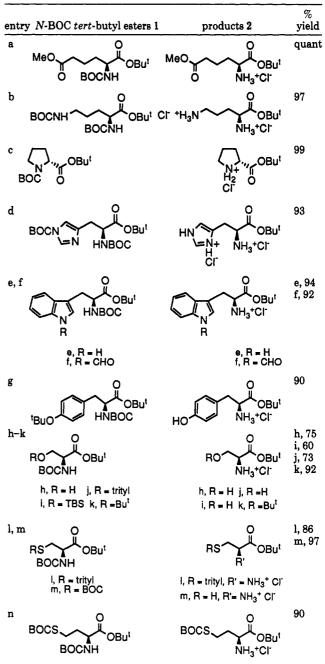
The selective removal of protecting groups is of critical importance in many synthetic sequences. The more selectively a protecting group can be removed, the more useful it becomes. We report here an efficient method for the selective removal of an N-BOC protecting group in the presence of a *tert*-butyl ester and several other common protecting groups.

During the course of earlier synthetic work,¹ it became necessary to remove an N-BOC while retaining a tertbutyl ester. Review of the literature revealed only one example of this transformation,² which we found to be slow and limited in scope. Examples of rapid N-BOC removal with dry HCl in ethyl acetate³ are known, and we considered that this might give the desired selectivity. Treatment of N-BOC-aminoadipic acid tert-butyl ester derivative 1a with a large excess of dry HCl in EtOAc for 3 h did indeed give the selective deprotection desired in 97% yield.¹ Further investigation established that as little as 500 mol % of a 1 M HCl in EtOAc solution would efficiently remove an N-BOC within 5 h at rt. In most cases, the product hydrochloride precipitated during the course of the reaction, possibly helping limit tert-butyl ester removal. The deprotection was carried out on a variety of N-BOC amino acid tert-butyl esters and derivatives, as shown in Table 1.

Each tert-butyl ester N-BOC amino acid derivative was treated under the same standard conditions, namely it was dissolved in a dry 1 M solution of HCl in EtOAc containing 500 mol % of HCl. The reaction mixture was stirred at rt until TLC indicated complete consumption of N-BOC tert-butyl ester 1. S- and N-BOC protected cysteine and homocysteine tert-butyl esters 1m and 1n were each treated with 1000 mol % of HCl to affect efficient cleavage of both the S- and N-BOC groups. In each case, amine hydrochloride product 2 was isolated either by filtration or after evaporation of the reaction mixture. Product hydrochlorides were usually analytically pure⁴ and could be used without further purification.

Several results point out additional selectivity for this method. *tert*-Butyl ethers tyrosine 1g and serine 1k demonstrated the greater acid sensitivity of an aryl *tert*butyl ether versus an alkyl *tert*-butyl ether. While the phenolic ether in tyrosine 1g was completely cleaved within

Table 1. Amino Acid tert-Butyl Ester Hydrochlorides 2 from N-BOC Substrates 1



2 h, the primary alcohol *tert*-butyl ether of serine 1k survived almost quantitatively. The acid lability of the S-BOC group demonstrated with entry 1m shows the BOC to be a versatile thiol protecting group, easily removed under basic or acidic conditions.³ Loss of the O-trityl from serine entry 1j was disappointing in light of the stability of the *tert*-butyl ether in 1k. The S-trityl group on cysteine derivative 11 did survive the deprotection conditions, contrasting differences between ether and thioether reactivity. We had hoped for some selectivity between the aryl and aliphatic N-BOC groups of histidine entry 1d, but both were rapidly removed with only 500 mol % of HCl to give crystalline *tert*-butyl ester dihydrochloride

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 (4) Products were greater than 95% pure as judged by NMR spectroscopy, and typically gave correct C, H, and N analyses upon drying.

product 2d. Our investigation reveals that subtle differences in the stability of some common protecting groups toward acid cleavage can be used to synthetic advantage.

Experimental Section

Procedures for the Synthesis of N-BOC Amino Acid tert-Butyl Esters. In all examples, N-BOC protecting groups were introduced by the method of Ponnusamy.⁵ All tert-butyl esters except for entries 1g and 1k were prepared using N,N'-diisopropyl-O-tert-butylisourea.¹ N'-Formyltryptophan was obtained using the method of Ohno.⁶ N-CBZ-serine and L-tyrosine were obtained by the method of Bergmann.⁷ O-Tritylserine was prepared by the method of Zervas.⁸ S-Tritylcysteine was synthesized according to Zervas.⁸ L-Homocysteine was produced from L-methionine by the method of du Vigneaud.⁹ NMR spectra were recorded at 300 MHz in CDCl₃ and J values are given in hertz.

N,N'-Bis-BOC-ornithine tert-butyl ester (1b): yield, 81% from ornithine; mp 80-82 °C; $[\alpha]^{25}_{D}$ +11.3° (c 1.3, CHCl₃); ¹H NMR δ 5.07 (br d, 1H, J = 7.8), 4.63 (m, 1H), 4.16 (t, 1H, J = 6.9), 3.14 (t, 2H, J = 6.0), 1.82-1.50 (m, 4H), 1.45 (s, 9H), 1.43 (s, 18H). Anal. Calcd for C₁₉H₃₆N₂O₆: C, 58.7; H, 9.3; N, 7.2. Found: C, 58.7; H, 9.7; N, 7.2.

N-BOC-proline tert-butyl ester (1c): yield, 81% from proline; $[\alpha]^{25}_{D}$ -50.5° (c 3.4, CHCl₃); ¹H NMR δ 4.12 (dd, 1H, J = 3.4, 8.9), 3.60–3.39 (m, 2H), 2.29–2.16 (m, 1H), 2.00–1.80 (m, 3H), 1.51 (s, 9H), 1.46 (s, 9H). Anal. Calcd for C₁₄H₂₆NO₄: C, 61.7; H, 9.6; N, 5.1. Found: C, 61.7; H, 9.6; N, 5.4.

N°,N°, Pis-BOC-histidine tert-butylester (1d): yield, 53% from histidine; mp 97–98 °C; $[\alpha]^{25}_{D}$ +17.8° (c 1.3, CHCl₃); IR 3880, 1750, 1710 cm⁻¹; NMR δ 7.98 (d, 1H, J = 0.8), 7.13 (s, 1H), 5.59 (br d, 1H, J = 8.4), 4.43 (dd, 1H, J = 2.9, 5.1), 3.00 (d, 2H, J = 5.2). Anal. Calcd for C₂₀H₃₄N₃O₆: C, 58.2; H, 8.3; N, 10.1. Found: C, 58.6; H, 8.2; N, 10.1.

(±)-N°-BOC-tryptophan tert-butyl ester (1e): yield, 70% from tryptophan; mp 188–189 °C; ¹H NMR δ 7.49 (d, 1H, J = 7.8), 7.32 (d, 1H, J = 8.0), 7.07–7.01 (m, 2H), 6.99–6.96 (m, 1H), 4.08–4.05 (m, 1H), 3.10–3.03 (m, 1H), 2.98–2.92 (m, 1H), 1.32 (s, 18H). Anal. Calcd for C₂₀H₂₈N₂O₄: C, 66.6; H, 7.8, N, 7.8. Found: C, 66.9; H, 7.8; N, 7.9.

(±)-N[·]-Formyl-N^{α}-BOC-tryptophan tert-butyl ester (1f): yield, 81% from (±)-N[·]-formyltryptophan;⁷ ¹H NMR δ 9.04 (s, 1H), 7.67–7.55 (m, 2H), 7.40–7.31 (m, 3H), 5.19–5.16 (m, 1H), 4.57–4.54 (m, 1H), 3.25–3.10 (m, 2H), 1.43 (s, 9H), 1.38 (s, 9H). Anal. Calcd for C₂₁H₂₈N₂O₅: C, 64.9; H, 7.3; N, 7.2. Found: C, 64.5; H, 7.4; N, 6.8.

N-BOC-*O***-***tert***-butyltyrosine** *tert***-butyl ester (1g)**: yield, 42% from *N*-CBZ-tyrosine, by the method used for the synthesis of 1k; $[\alpha]^{25}_{D} + 37.3^{\circ}$ (c 1.5, CHCl₃); ¹H NMR δ 7.10–6.89 (m, 4H), 5.01 (br d, 1H, J = 7.9), 4.43–4.40 (m, 1H), 2.99 (d, 2H, J = 6.1), 1.42 (s, 9H), 1.39 (s, 9H), 1.35 (s, 9H). Anal. Calcd for C₂₂H₃₅-NO₅: C, 67.1; H, 9.0; N, 3.6. Found: C, 67.0; H, 9.0; N, 3.6.

N-BOC-serine tert-butylester (1h): yield, 56% from serine; mp 76-78 °C; $[\alpha]^{25}_D$ -20° (c 1.8, EtOH); ¹H NMR δ 5.40 (br s, 1H), 4.20 (br s, 1H), 3.84 (m, 2H), 2.55 (m, 1H), 1.45 (s, 9H), 1.40 (s, 9H). Anal. Calcd for C₁₂H₂₃NO₅: C, 55.2; H, 8.9; N, 5.6. Found: C, 55.2; H, 9.3; N, 5.6.

N-BOC-O-tert-butylserine tert-Butyl Ester (1k). N-CBZserine (5.0 g, 25 mmol) was dissolved in 100 mL of dry dichloromethane. The solution was cooled to 0 °C and saturated with isobutylene, after which 1 mL of sulfuric acid was added and the reaction mixture was sealed at 0 °C and stirred at rt for 4 days. The reaction vessel was then carefully depressurized, and the mixture was washed with two 50-mL portions of saturated aqueous NaHCO₃ solution, dried, and evaporated. The residue of crude N-CBZ-O-tert-butylserine tert-butyl ester (6.73 g) was dissolved in dry methanol (50 mL), and to this solution were added BOC₂O (5 g, 23 mmol) and 0.6 g of 5% Pd/C. The mixture

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was stirred under an atmosphere of H₂ (balloon) overnight, after which it was filtered through Celite, evaporated, and chromatographed to give 1k as a clear oil: 67% from L-serine; $[\alpha]^{25}_{D}+5.0^{\circ}$ (c 1.4, CHCl₃); ¹H NMR δ 5.32 (br d, 1H, J = 8.8), 4.25 (dt, 1H, J = 8.7, 2.6), 3.76, (dd, 1H, J = 8.6, 2.7), 3.51 (dd, 1H, J = 8.6, 2.9), 1.58–1.18 (m, 27H). Anal. Calcd for C₁₆H₃₁NO₅: C, 60.5; H, 9.8; N, 4.4. Found: C, 60.3; H, 9.6; N, 4.1.

N-BOC-S-tritylcysteine tert-butyl ester (11): yield, 46% from cysteine; $[\alpha]^{25}_{D} + 12.9^{\circ}$ (c 2.6, CHCl₃); ¹H NMR δ 7.43–7.19 (m, 15H), 5.10 (d, 1H, J = 8.1), 4.22–4.19 (m, 1H), 2.52 (d, 1H, J = 4.5), 1.46 (s, 9H), 1.43 (s, 9H). Anal. Calcd for C₃₁H₃₇NO₄S: C, 71.6; H, 7.1; N, 2.7. Found: C, 72.0; H, 7.1; N, 2.8.

N,S-Bis-BOC-cysteine tert-butyl ester (1m): yield, 71% from cysteine; $[\alpha]^{25}_{D}$ +12.8° (c 1.7, CHCl₃); ¹H NMR δ 5.29 (d, 1H, J = 7.5), 4.45 (m, 1H), 3.20 (dd, 1H, J = 5.7, 14.5), 2.95 (dd, 1H, J = 3.9, 8.7), 1.50–1.41 (m, 27H). Anal. Calcd for C₁₇H₂₂NO₆S: C, 54.0; H, 8.2; N, 3.7. Found: C, 54.2; H, 8.1; N, 3.6.

N,S-Bis-BOC-homocysteine tert-butyl ester (1n): yield, 66% from homocysteine;¹⁰ $[\alpha]^{25}_{D}$ +6.6° (c 0.9, CDCl₃); ¹H NMR δ 5.11 (br d, 1H, J = 4.0), 4.43-4.20 (m, 1H), 2.84-2.75 (m, 2H), 2.14-1.89 (m, 2H), 1.60-1.40 (m, 27H). Anal. Calcd for C₁₈H₃₃NO₆S: C, 55.1; H, 8.4; N, 3.5. Found: C, 55.4; H, 8.2; N, 3.6.

General Procedure for the Selective Removal of an N-Boc Group. The N-BOC amino acid ester (1 mmol) was dissolved in 500 mol % of a 1 M solution of HCl in ethyl acetate (prepared by bubbling dry HCl into dry ethyl acetate then diluting to 1 M with additional ethyl acetate). The reaction mixture was stirred at room temperature until the disappearance of starting material as determined by TLC (typically 3-5 h). At this time the precipitated product was isolated by filtration, or in the case of soluble products, the reaction mixture was evaporated in the cold and the residue crystallized by trituration with anhydrous ether, followed by filtration to isolate the product. In all cases, the crude isolated solid was greater than 95% pure, as determined by ¹H NMR in D₂O. All NMR spectra were recorded at 300 MHz in D₂O unless otherwise specified.

Ornithine *tert*-butyl ester dihydrochloride (2b): yield, 97% from 1b; mp 195–196 °C; $[\alpha]^{25}_{D}$ +7.4° (*c* 1.4, H₂O); ¹H NMR δ 4.05 (t, 1H, J = 6.3), 3.05 (t, 2H, J = 7.5), 2.10–1.75 (m, 4H), 1.53 (s, 9H). Anal. Calcd for C₉H₂₂N₂Cl₂O₂: C, 41.3; H, 8.9; N, 10.7. Found: C, 41.1; H, 8.5; N, 10.7.

Proline tert-butyl ester hydrochloride (2c): yield, 99% from 1c; mp 108–109 °C (lit.¹⁰ mp 109–111 °C); $[\alpha]^{25}_{D}$ –31.6 (c 1.2, EtOH) [lit.¹⁰ $[\alpha]$ –31.0° (c 2, EtOH)]; ¹H NMR δ 4.40–4.30 (m, 1H), 3.50–3.34 (m, 2H), 2.50–2.36 (m, 1H), 2.21–2.00 (m, 3H), 1.50 (s, 9H).

Histidine tert-butylester dihydrochloride (2d): yield, 93% from 1d; mp 90 °C dec; $[\alpha]^{25}_D$ +19.1° (c 1.7, D₂O); ¹H NMR (DMSO) δ 8.51 (s, 1H), 7.20 (s, 1H), 4.13 (t, 1H, J = 6.2), 3.18 (d, 2H, J = 7.4), 1.18 (s, 9H). Anal. Calcd for C₁₅H₂₈N₃ClO₄: C, 42.3; H, 6.7; N, 14.7. Found: C, 42.6; H, 6.6; N, 14.2.

(±)-Tryptophan tert-butyl ester hydrochloride (2e): yield, 94% from 1e; mp 212–213 °C dec [lit.¹⁰ mp 214 °C dec]; ¹H NMR δ 7.51 (d, 1H, J = 8.0), 7.38 (d, 1H, J = 8.1), 7.15–7.11 (m, 2H), 7.04, (t, 1H, J = 7.4), 4.16 (t, 1H, J = 6.8), 3.28 (t, 2H, J = 7.0), 1.22 (s, 9H).

(±)-N-Formyltryptophan tert-butyl ester hydrochloride (2f): yield, 92% from 1f; mp 164–165 °C dec; ¹H NMR δ 8.86 (br s, 1H), 7.48–7.45 (m, 2H), 7.31–7.26 (m, 3H), 4.20 (t, 1H, J = 7.1), 3.17 (d, 2H, J = 7.1), 1.17 (s, 9H). Anal. Calcd for C₁₆H₂₁N₂O₃·H₂O: C, 57.6; H, 6.6; N, 8.4. Found: C, 57.6; H, 6.3; N, 8.8.

Tyrosine *tert*-butyl ester hydrochloride (2g): yield, 90% from 1g; mp >200 °C; $[\alpha]^{25}_{D}$ +4.5° (*c* 1.2, H₂O); ¹H NMR δ 7.01– 6.73 (m, 4H), 4.02 (t, 1H, *J* = 6.7), 2.98 (d, 2H, *J* = 6.7), 1.24 (s, 9H). Anal. Calcd for C₁₃H₂₀NClO₃: C, 57.0; H, 7.3; N, 5.1. Found: C, 57.0; H, 7.4; N, 5.2.

Serine tert-butyl ester hydrochloride (2h): yield, 75% from 1h; mp 111–113 °C; $[\alpha]^{25}_D$ -4.3° (c 1.8, EtOH); ¹H NMR δ 3.99–3.97 (m, 1H), 3.93 (dd, 1H, J = 12.4, 4.5), 3.84 (dd, 1H,

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J = 12.3, 3.4), 1.31 (s, 9H). Anal. Calcd for C₇H₁₆NO₃·0.5H₂O: C, 40.7; H, 8.3; N, 6.8. Found: C, 40.9; H, 7.9; N, 7.2.

O-tert-Butylserine tert-butyl ester hydrochloride (2k): yield, 85% from 1k; mp 149–150 °C; $[\alpha]^{25}_{D}$ –8.7° (c 1.4, H₂O); 'H NMR δ 4.16 (dd, 1H, J = 3.4, 4.3), 3.90 (dd, 1H, J = 4.5, 10.4), 3.78 (dd, 1H, J = 3.4, 10.4), 1.55 (s, 9H), 1.20 (s, 9H). Anal. Calcd for C₁₁H₂₄NClO₃: C, 52.0; H, 9.5; N, 5.5. Found: C, 51.6; H, 9.2; N, 5.5.

S-Tritylcysteine tert-butyl ester hydrochloride (21): yield, 86% from 11; characterized as the N-acetyl derivative. Compound 21 (0.1 g, 0.2 mmol) was dissolved in 2 mL of pyridine. Acetic anhydride (0.5 mL) was added and the reaction mixture was stirred at rt for 2 h. The mixture was concentrated and the residue chromatographed to give N-acetyl-S-tritylcysteine tert-butyl ester: 0.079 g, 78%; mp 48-50 °C; [α]²⁵_D+6.0° (c 1.4, CHCl₃); ¹H NMR (CDCl₃) δ 7.46-7.20 (m, 15H), 5.93 (d, 1H, J = 7.7), 4.54 (m, 1H), 2.64 (dd, 1H, J = 5.4, 12.0), 2.49 (dd, 1H, J = 4.6, 12.0), 1.98 (s, 3H), 1.46 (s, 9H). Anal. Calcd for C₂₈H₃₁NO₃S: C, 72.8; H, 6.7; N, 3.0. Found: 72.6, 6.9, 3.1.

Cysteine tert-butyl ester hydrochloride (2m): yield, 94% from 1m; mp 145–147 °C; $[\alpha]^{25}D_{-17.2°}$ (c 0.9, D₂O): ¹H NMR δ 4.30 (m, 1H), 3.20–3.00 (m, 2H), 1.52 (s, 9H). Anal. Calcd for C₇H₁₆NClO₂S: C, 39.3; H, 7.5; N, 6.5. Found: C, 39.4; H, 7.1; N, 6.3.

S-BOC-Homocysteine tert-butyl ester hydrochloride (2n): yield, 90% from 1n; characterized as N-acetyl-S-BOChomocysteine tert-butyl ester prepared by the procedure for 21: yield, 63% from 2n; $[\alpha]^{25}_{D}$ +18.4° (c 0.7, CDCl₃); ¹H NMR δ 6.15 (d, 1H, J = 7.5), 4.51-4.44 (m, 1H), 2.15-2.04 (m, 1H), 1.98-1.85 (m, 1H), 1.44 (s, 9H), 1.41 (s, 9H). Anal. Calcd for C₁₅H₂₇NO₅S: C, 54.0; H, 8.1; N, 4.2. Found: C, 53.7; H, 7.9; N, 3.9.